

The first synthesis of a 2',3'-methano carbocyclic nucleoside¹

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Abstract

A 2',3'-methano carbocyclic nucleoside, 9-(c-4-hydroxymethylbicyclo[3.1.0]hex-<math>r-2-yl)-9H-adenine, has been efficiently synthesized from 2-azabicyclo[2.2.1]hex-5-en-3-one(ABH) in six steps involving 1,3-dipolar cycloaddition, photolysis, and adenine ring construction. © 1999 Elsevier Science Ltd. All rights reserved.

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A cyclopropane fused nucleoside, namely methano-nucleoside, has recently attract much attention because of its pharmacological and biological activities. A methano cytidine derivative (**A**) shows antiviral activity.² Three types of methano carbocyclic nucleosides (**B**,³ **C** and **D**) have been synthesized so far. Carbocyclic 4',6'-methano- (**C**) and 1',6'-methano-thymidine (**D**) assume 2'-exo and 3'-exo conformations, respectively. The nucleoside (**C**) shows significant anti-HSV activity⁴ and the oligomer containing **C** has been found to stabilized the duplex.⁵ The oligomer consisting of **D** also forms the duplex with the complemental natural oligomer.⁶ Quite recently, a cyclopentenyl guanine derivative (**E**: carbovir) has been approved as a drug for the treatment of AIDS.⁷ Previously, we reported an efficient synthesis of a precursor of abacavir from 2-azabicyclo[2.2.1]hept-5-en-3-one (ABH) via π-allylpalladium complex.⁸ We planned to design a methano derivative (**F**) because a cyclopropane ring as well as a C-C double bond has unsaturated property and therefore **F** can be expected to show anti-HIV activity. In this communication, we now report the first synthesis of racemic **F** from ABH.

Cyclopropane construction was carried out using diazomethane because Simmons-Smith reaction of ABH or its N- $^{\prime}$ Boc derivative (1) 9 did not give the desired compound. 10 When 1 was treated with an

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excess of diazomethane-ether solution¹¹ containing methanol at room temperature overnight, a mixture of 2^{12} and 3^{12} was obtained with the 1:1 ratio in 69% yield concomitant with the formation of 4 (20%). The structure of 2 was confirmed by NOE experiments as shown in Scheme 1. Since 4 did not react with diazomethane under the same conditions, compounds 2 and 3 would be formed by the 1,3-dipolar cycloaddition of 1 with diazomethane followed by solvolysis with methanol. The initial cycloaddition selectively occurred from the exo side of 1 in the same manner as the epoxidation of 1 by m-CPBA.¹³ It is also proved by the latter mentioned reaction. Formation of 4 might be assisted by diazomethane (Scheme 1) because conversion of 1 to 4 did not take place in methanol under neutral condition. 14 Reaction of 1 with diazomethane in the absence of methanol¹¹ gave a mixture of regioisomers 5¹⁵ and 6¹⁵ in 58% and 35% yields, respectively. Determination of the structure of 5 and 6 was made by their conversion to 2 and 3, respectively. Both compounds were readily converted to the corresponding bicyclo-compound 2 and 3 by methanolysis at room temperature for 10 min. Although 2 and 3 were chromatographically separable, the mixture was used for the following reaction without separation. Thus, a solution of the mixture in acetonitrile was irradiated with high pressure marcury lamp (Pyrex filter) with ice-cooling for 2 h to give a cyclopropane derivative (7) in 94% yield. Reduction of 7 with Ca(BH₄)₂ in ethanol afforded an alcohol (8)16 in 81% yield.

Scheme 1.

A more straightforward synthesis of **8** was examined. Compound (**1**) was treated with diazomethane-ether solution without methanol in the presence of PdCl₂(PhCN)₂ to give directly a tricyclic compound (**9**) in 41% yield, which was consistent with the compound obtained from the photolysis of **5** and **6** in 42% and 60% respective yields. Compound (**9**) was then submitted to reductive amido cleavage (RAC) reaction using NaBH₄ developed previously by our group¹⁷ to form **8** in 70% yield. Cyclopropanation of **4** with diazomethane using transition metal catalysts was also investigated to obtain **7** directly. Use of PdCl₂(PhCN)₂ resulted in the formation of **7** in low yield (23%) whereas Rh₂(OAc)₂ did not catalyze the reaction. Removal of the 'Boc group of **8** with trifluoroacetic acid (TFA) gave the desired product (**10**), which, without purification, was submitted to the manipulation of adenine ring construction to give the final product (**13**)¹⁸ via **11** and **12** as shown in Scheme 2.

In conclusion, we have achieved the first synthesis of 2',3'-methano carbocyclic nucleoside (13) from ABH in ca. 12% total yield. We have found that 1 is more reactive than the cyclopentene (4) for both the

Scheme 2. (a) 5-Amino-4,6-dichloropyrimidine, *i*-Pr₂NEt, *n*-BuOH, reflux, 24 h; (b) CH(OEt)₃, 10N HCl, rt, 20 h; (c) NH₃, MeOH, 50°C, 24 h (sealed tube)

1,3-dipolar cycloaddition with diazomethane and the cyclopropanation using $PdCl_2(PhCN)_2$. This might be attributable to factor 'X'¹⁹ or staggering of allylic bond²⁰ as observed in the cycloaddition to the C=C bond of norbornene. Works on the antiviral evaluation of 13 and the synthesis of its guanine congener are in progress, and the results will be reported in due course.

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References

- 1. Part ILV of Synthesis of nucleosides and their related compounds. For Part ILIV, see: Katagiri, N.; Morishita, Y.; Oosawa, U.; Yamaguchi, M. *Tetrahedron Lett.* **1999**, *40*, 6835–6840.
- 2. Okabe, M.; Sun, R.-C. Tetrahedron Lett. 1989, 30, 2203-2206.
- 3. Chang, H. S.; Bergmeier, S. C.; Frick, J. A.; Bathe, A.; Rapoport, H. J. Org. Chem. 1994, 59, 5336-5342.
- Marquez, V. E.; Siddiqui, M. A.; Ezzitouni, A.; Russ, P.; Wang, J.; Wagner, R. W.; Matteucci, M. D. J. Med. Chem. 1996, 39, 3739-3747.
- 5. Altmann, K.-H.; Kesselring, R.; Francotte, E.; Rihs, G. Tetrahedron Lett. 1994, 35, 2331-2334.
- 6. Altmann, K.-H.; Imwinkelried, R.; Kesselring, R.; Rihs, G. Tetrahedron Lett. 1994, 35, 7625–7628.
- 7. (a) Vince, R.; Hua, M. J. Med. Chem. 1990, 33, 17-21. (b) Vince, R.; Brownell, J. Biochem. Biophys. Res. Commun. 1990, 168, 912 and references cited therein.
- 8. Katagiri, N.; Takebayashi, M.; Kokufuda, H.; Kaneko, C.; Kanehira, K.; Torihara, M. J. Org. Chem. 1997, 62, 1580-1581.
- 9. Toyota, A.; Habutani, C.; Katagiri, N.; Kaneko, C. Tetrahedron Lett. 1994, 35, 5665-5668.
- 10. Unpublished results.
- 11. An etheral solution of diazomethane containing methanol was obtained from the treatment of *N*-methyl-*N*-nitroso-*p*-toluenesulfonylamide with sodium hydroxide in methanol–water whereas methanol free diazomethane–ether solution was prepared using carbitol instead of methanol.
- 12. Compound **2** (less polar): mp 123°C (dec.) (hexane–AcOEt). ¹H NMR (400 MHz, CDCl₃): 1.43 (9H, s), 1.80 (1H, dt, J=13.7, 8.4 Hz), 1.92 (1H, dt, J=13.7, 5.1 Hz), 2.40 (1H, br s), 3.35 (1H, br s), 3.58 (1H, br s), 3.82 (3H, s), 4.55 (1H, dd, J=18.6, 9.0 Hz), 4.70 (1H, dt, J=18.6, 2.9 Hz), 5.11 (1H, br s), 5.30 (1H, ddd, J=8.3, 5.4, 2.9 Hz). **3** (more polar): ¹H NMR (300 MHz, CDCl₃): 1.48 (9H, s), 1.92 (2H, m), 2.31 (1H, dd, J=6.3, 2.7 Hz), 2.83 (1H, m), 3.72 (3H, s), 4.39 (1H, dd, J=18.3, 8.4 Hz), 4.44 (1H, br s), 4.66 (1H, d, J=18.4 Hz), 4.93 (1H, br s), 5.41 (1H, br s).
- Katagiri, N.; Matsuhashi, Y.; Kokufuda, H.; Takebayashi, M.; Kaneko, C. Tetrahedron Lett. 1997, 38, 1645–1648. The exo-selectivity would be due to pyramidarization [Williams, R. V.; Gadgil, V. R.; Garner, G. G.; Williams, J. D.; Vij, S. Tetrahedron Lett. 1999, 40, 2689–2690, and references cited therein] or Cieplak effect [Cieplack, A. S.; Tait, B. D.; Johnson, C. R. J. Am. Chem. Soc. 1989, 111, 8447–8462] as observed in the selectivity for the addition to a double bond of bicyclic compound.
- 14. Compound 4 was prepared in 54% yield by refluxing of 1 in methanol with catalytic amount of conc. hydrochloric acid for 4 h, mp 36°C.

- 15. Compound 5 (less polar): mp 143°C. ¹H NMR (300 MHz, CDCl₃): 0.83 (1H, br d, J=11.3 Hz), 1.58 (9H, s), 1.85 (1H, dt, J=11.3, 1.5 Hz), 2.51–2.61 (1H, m), 2.63 (1H, br s), 4.23 (1H, dt, J=19.0, 3.4 Hz), 4.71 (1H, ddd, J=19.0, 9.7, 1.7 Hz), 5.06 (1H, t, J=1.5 Hz), 5.29–5.36 (1H, m). 6 (more polar): mp 148–150°C. ¹H NMR (300 MHz, CDCl₃): 0.83 (1H, br d, J=11.4 Hz), 1.53 (9H, s), 1.86 (1H, br d, J=11.4 Hz), 2.54–2.65 (1H, m), 3.47 (1H, br s), 4.23 (1H, dt, J=19.0, 4.2 Hz), 4.29 (1H, br s), 4.59 (1H, ddd, J=19.0, 9.8, 1.5 Hz), 5.28 (1H, br dd, J=7.1, 1.1 Hz).
- 16. ¹H NMR (400 MHz, CDCl₃): 0.00 (1H, m, overlapped with TMS), 0.49 (1H, dt, J=8.3, 5.4 Hz), 1.34 (2H, m), 1.44 (10H, s and m), 1.78 (1H, dt, J=14.6, 7.3 Hz), 2.16 (1H, dt, J=8.8, 4.4 Hz), 2.80 (1H, br), 3.58 (1H, dd, J=10.3, 4.4 Hz), 3.70 (1H, dd, J=10.3, 3.9 Hz), 4.00 (1H, br), 5.90 (1H, d, NH).
- 17. Katagiri, N.; Muto, M.; Kaneko, C. Tetrahedron Lett. 1989, 30, 1645-1648.
- 18. Compound 13: mp 215–216°C. ¹H NMR (400 MHz, CD₃OD): 0.44 (1H, dt, J=4.9, 3.9 Hz), 0.88 (1H, dt, J=5.3, 8.4 Hz), 1.81 (1H, dt, J=8.3, 4.4 Hz), 1.94 (1H, dt, J=8.3, 3.9 Hz), 2.07 (1H, dt, J=15.6, 7.8 Hz), 2.28 (1H, q, J=7.8 Hz), 3.18 (1H, dd, J=10.8, 7.8 Hz), 3.37 (1H, dd, J=10.8, 7.3 Hz), 4.91 (1H, d, J=6.8 Hz), 8.20 (1H, s), 8.36 (1H, s). Anal. calcd for C_{1.7}H₁₃N₅O: C, 59.25; H, 5.39; N, 28.79. Found: C, 59.15; H, 5.50; N, 28.63.
- 19. Huisgen, R.; Ooms, P. H. J.; Mingin, M.; Allinger, N. L. J. Am. Chem. Soc. 1980, 102, 3951-3953.
- 20. Rondan, N. G.; Paddon-Row, M. N.; Caramella, P.; Mareda, J.; Mueller, P. H.; Houk, K. N. J. Am. Chem. Soc. 1982, 104, 4974-4976.